Coordination Chemistry Reviews, 53 (1984) 296
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Errata

B. Corain, The Coordination Chemistry of Hydrogen Cyanide, Cyanogen and Cyanogen Halides, Coord. Chem. Rev., 47 (1982) 165–200.

p. 170, Table 4, second line, should read:

NC–CN $D_{\infty h}$ $\text{C–C} = 1.37 \pm 0.02$; $\text{C}\equiv\text{N} = 1.16 \pm 0.02$

Structure III does not contribute significantly and should be deleted. The sentence on p. 170 beginning: 'There is no doubt...,' should read:

There is no doubt that a contribution from such a ground state would account for the particularly high dissociation energy of the C–C bond in C_2N_2, as well as the short C–C bond distance, 1.37 ± 0.02 Å [the sum of two C(*sp*) covalent radii is 1.40 Å].

A.B.P. Lever, Analytical Treatment of the Electronic Spectra of Some Low Symmetry Transition Metal Complexes, Coord. Chem. Rev., 43 (1982) 63–85.

Equations (15)–(21) for the square pyramidal $ML_4Z^+ Z^-$ species are only appropriate for the case where $e_\pi(L) = 0$.

In eqn. (A4), the ratio (1/990) should qualify the entire expression not simply the first term as presently indicated.

Eqn. (A9), is incorrect and should read:

$$\begin{aligned} DT &= -(7\sqrt{15}\,Dt)/2 \\ &= -(\sqrt{15}/20)\left[11\nu_3 - 3\nu_4 - (\nu_1 + \nu_5) - 2(\nu_2 + \nu_6) + 45B\right] \end{aligned}$$

Thus the expression for *DT* does depend upon the d^7 ground state.